Highly Enantioselective Creation of Quaternary Carbons in a Halolactonization of Bis-γ,δ-unsaturated Carboxylic Imides Derived from a Camphorsultam: **Enantioselective Synthesis of (+)-Mesembrine**

Tsutomu Yokomatsu, Haruo Iwasawa, and Shiroshi Shibuya*

Tokyo College of Pharmacy, 1432-1 Horinouchi, Hachioji, Tokyo 192-03, Japan

Abstract: Iodolactonization of symmetrical diene-carboxylic imides **(3a,b),** derived from a camphorsultam, afforded chiral lactones (5) and (6) possessing aromatic substituents on quaternary carbons in high enantioselectivity with moderate diastereoselectivity.

Creation of asymmetric quatemary carbon center is one of the most challenging problems in synthetic organic chemistry.¹ A number of methods have been reported recently for the highly enantioselective construction of quaternary carbon centers in various molecular frames.² Enantioselective halolactonization of symmetrical diene-carboxylic chiral amides (1a,b) having quaternary carbons through discrimination of diastereotopic alkene group with concomitant face differentiation was also pursued in order to obtain asymmetric synthons with diastereo- and enantioisomeric purity.³ However it was rather difficult to achieve high enantioselectivity.^{3,4} Recently we have found that iodolactonization of symmetrical diene-carboxylic imide (2) derived from a camphorsultam, proceeded in high degree of diastereo- and enantioselectivity.⁵ In this paper, we report our finding on the iodolactonization of symmetrical diene-carboxylic imides (3) possessing aromatic substituents on quatemary carbons; this cyclization affords the chiral lactones in high yield as well as high enantioselectivity.

Symmetrical diene-carboxylic acid **(4a),** prepared from ethyl 3.4-dimethoxyphenylacetate through sequential diallylation and saponification, was condensed with a sultam, derived from D-camphorsulfonic acid, under the methods⁶ reported by Oppolzer, to give N-acylsultam $(3a)$,⁷ mp 151-153 °C $[\alpha]_D^{20}$ –55.6° (c 1.1, CHCl₃), in 65% yield. Treatment of 3a with iodonium di-sym-collidine perchlorate (2.3 equiv.) in CH₂Cl₂-

MeOH containg 1.2 equiv. of H₂O at -50 °C for 48 h gave a diastereoisomeric mixture of 5a,⁸ [α]_D²⁰+43.0° (c 0.5, CHCl₃), and 6a in a ratio of $4.5:1$ in 83% yield. The relative stereochemistry of 5a and 6a was established by NOE experiments on 5a. Representative NOE data for 5a are shown in A (Scheme 1). The enantiomeric excess (e.e.) of 5a was determined to be 88% by ¹H NMR (300 MHz) analysis with (S)-(+)-9anthryl trifuoromethylcarbinol as chiral solvating agent.⁹

Scheme 1

In order to ascertain absolute stereochemistry of the quaternary carbon centers in both diastereoisomers, deiodomethylations of 5a and 6a were carried out. Treatment of 5a with aqueous potasium hydroxide (20%), followed by acid work-up (HCl) afforded γ -lactone (7a) quantatively as a mixture of diastereoisomers. The compound (7a), on sequential saponification, periodate cleavage and reduction in MeOH in the same flask [1. 10% KOH then CO₂, 2. NaIO₄, 3. NaBH₄], gave the lactone $(8a)$, $(α]_D^{20}+74.1°$ (c 0.5, MeOH), in 80% yield. Since 6a could not be isolated as pure states, similar transformations as above were carried out on an 1.6 : 1 mixture of 5a and 6a to give 8a. The sigh of specific rotation and e.e. (88%)¹⁰ of which were identical those from pure 5a. Then, absolute stereochemistry on the quaternary carbons of 6a was found to be same as that of 5a. Thus for the synthesis of 8a, a mixture of 5a and 6a was used, without separation, practically.

In the efforts to deduce absolute configuration, 8a was converted to keto lactone 11a. The absolute stereochemistry of (-)-11a has been already established by Takano and co-workers through chemical correlation

to natural (-)-mesembrine.¹¹ Ozonolysis of a_1 [1. O₃, pyridine-CH₂Cl₂, -78 °C; 2. Me₂S] gave aldehyde $9a$, $[\alpha]_D^{20}+73.8^\circ$ (c 1.04, CHCl3), as an oil in 70% yield. Wittig reaction of 9a [MeCH=PPh3, THF], followed by palladium catalyzed oxidation¹² [PdCl₂, CuCl, O₂, wet DMF] gave the requsite methyl ketone 11a, an oil, $[\alpha]_0^{20}$ +69.1° (c 0.44, MeOH); lit.,¹¹ mp 78 °C, $[\alpha]_0^{20}$ -42.6° (c 0.5, MeOH) for (-)-11a. Observed optical rotation of 11a revealed that it was antipodal to that reported in the literature.¹¹ Thus, the stereochemistry of 5a and 6a was unambiguously assighed as shown in **Scheme 1.**

Iodolactonization of 3b14 **under** the same **conditions as for** 3a **proceeded similarly in a high** enantioselectivity to give a diastereoisomeric mixture of 5b, 84% ee,¹³ [α]_D²⁰+50.1° (c 1.05, CHCl₃), and 6b $(5b : 6b = 5.5 : 1, 65\%$ yield). A mixture of 5b and 6b was converted to 8b, $[\alpha]_D^{20+53.7^{\circ}}$ (c 0.3, MeOH), *via* 7b **through exactly the same proce@re as for 8a. It should be noted that aromatic substituents on quaternary carbons of 3 wen quite important to induce the halolactonization with high enantioselectivity.** Upon **treatment of 3~14 possessing alky! substituent such as ethyl with iodonium di-sym-collidine** perchlorate under identical conditions to that used for 3a. diasteroisomeric mixture (1 : 1) of 1215 was obtained **in 75% yield. Optical purity** of 13 derived from 12, however, appeared to be very low($25\%, ^{4,16}$ [13: $\left[\alpha\right]_D$ ²⁰ -3.28° (c 0.36, MeOH), lit.,⁴ $[\alpha]_D^{20}+12.77^{\circ}$ (c 3.1, MeOH) for (+)-13].

Although precise stereochemical course of the present iodolactonization is unclear at this stage, the results delineated in this paper strongly suggest that the observed enantioselection may not reflect on simple steric interaction alone.⁵ Assuming two transition states A and B leading to 5a and its enantiomer, 17 transition state A would be favored over B , because transition states A involves π -staking stabilization between sulfone function and aromatic ring but in transition state B steric repulsion between sulfone oxygene and ally1 substituent are presence. A clear understanding of the stereoselectivity in these halolactonization must await further experimentation.

In summary we have demonstrated a usefulness of iodolactonization of symmetrically diene-carboxylic imides possessing a sultam as chiral auxiliary to create quaternary carbon centers in high enantioselectivity. The results described in this paper also constitute enantioselective synthesis of $(+)$ -mesembrine in a formal manner.¹¹

REFERENCES AND NOTES

- 1. For reviews: see a) Martin, S. F. Tetrahedron 1980, 36, 419. b) ApSimon, J. W.; Collier, T. L. Ibid. 1986, 42, 5157. c) Asymmetric Synthesis; Morrison, J. D., Ed.; Academic Press: New York, 1983.
- 2. a) Takano, S.; Shima2aki.Y.; Moriya. M, Ogasawara, K. Chem. Lert. 1990, 1177. b) Isaka, M; Nakamura, E. J. Am. Chem. Soc. 1990, 112, 7428. c) Lee, E.; Shi, I-J.; Kim, T-S. J. Am. Chem. Soc. 1990,112,260 and references cited in therein.
- 3. Hart, D. J.; Huang, H.-C.; Krishnamurthy, R.; Schwartz, T. J. Am. Chem. Soc. 1989, 111, 7507.
- 4. Takano, S.; Murataka, C.; Imamura, Y.; Tamura, N.; Ogasawara, K. *Heterocycles* 1981, 16, 1291.
- 5. Yokomatsu. T.; Iwasawa, H.; Shibuya, S. *J. Chem. Sot.,* Chem. Commun., 1992, 728. Highly enantioselective halolaconization of symmettical diene acids using (2R,5f?)-bis(methoxymethyl)pyrrolidine as a chiral auxiliary was first reported by Fuji et al.; Fuji, K.; Node, M.; Naniwa, Y.; Kawabata, T. *Tetrahedron Left. 1990.31, 3 175.*
- 6. For review: Oppolzer, W. Tetrahedron 1987,43, 1969.
- 7. All new compounds gave satisfactory spectroscopic and analytical data.
- 8. 5a: an oil; ¹H NMR (300 MHz, CDCl₃, δ) 6.97 (1H, d, J=2.2 Hz), 6.91 (1H, dd, J=2.2, 8.5 Hz), 6.84 (lH, d, J=8.5 Hz), 5.71-5.58 (lH, m), 5.17-5.08 (2H, m), 4.30 (lH, dddd, J=4.8, 5.1, 7.5. 10.5 Hz), 3.90 (3H, s), 3.89 (3H, s), 3.43 (lH, dd. J=4.8, 10.3 Hz), 3.26 (lH, dd. J=7.5. 10.3 Hz), 2.88 (1H, dd, J=5.1, 13.2 Hz), 2.66 (2H, d, J=7.1 Hz), 2.12 (1H, dd, J=10.5, 13.2 Hz). 8a: an oil: ¹H NMR (300 MHz, CDC13, 6) 7.05 (IH, d, J=2.3 Hz), 6.93 (1H. dd. J=2.3, 8.5 Hz), 6.83 (1H. d, J=8.5 Hz), 5.67-5.56 (lH, m), 5.14-5.06 (2H, m), 4.31 (1H. ddd, J=3.2, 8.9, 12.3 Hz), 4.12 (lH, ddd, J=6.4, 9.3, 12.3 Hz) 3.89 (3H, s), 3.87 (3H, s), 2.69-2.43 (4H, m).
- 9. Pirkle, W. H.; Sikkenga, D. J. J. *Org. Chem. 1977,42, 1370.*
- 10. Determined by $\rm{^{1}H}$ NMR(300 MHz) with Pirkle agent.⁹
- 11. Takano, S.; Imamura, Y.; Ogasawara, K. *Tetrahedron Lett.* 1981, 22, 447
- 12. Tsuji, J.; Shimizu, I.; Yamamoto. K. *Tetrahedron Left. 1976.2957.*
- 13. Determined by lH NMR(300 MHz) with Eu(hfc)3.
- 14. 3b: mp 141-142 °C, $[\alpha]_D^{20}$ -47.9° (c 1.0, CHCl3). 3c: mp 57-59 °C, $[\alpha]_D^{20}$ +3.29° (c 1.0, CHCl3).
- 15. Because it was difficult to separate individual stereoisomer of 12. we could not determine the optical purities for each diastereoisomers; optical purity with one of diastereoisomers of the iodolactones, prepared by analogous iodolactonization of 3 (R=Me), could be directly determined as 25% e.e.¹³
- 16. Absolute configuration and optical purity of 13 were determined by comparison of the optical rotation value with that in the literature. 4
- 17. In order to explain the observed diastereoselectivity (aromatic substituent rruns to iodomethyl **group),** aromatic substituent is assumed to be as pseudoaxial configuration in the transition states.